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state,  $N_{\text{high}}/N_{\text{low}}$ , for  $^{13}\text{C}$  at  $25^\circ\text{C}$  in an applied  
field strength of  $7.05\text{ T}$ .

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Chapter (c)  $\text{C-Cl} + \text{Cl-C}$  Chlorine is  
more electronegative than carbon

Problem 1.6 Draw Lewis structures,  
showing all valence electrons, for these

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